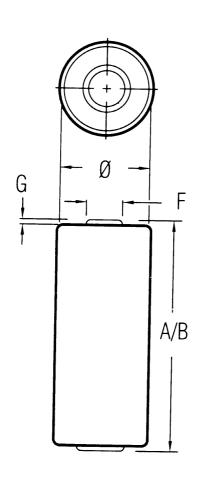


1.5V IEC LR1

LR1

No Added Mercury or Cadmium Chemical System: Alkaline-Manganese Dioxide (Zn/Mn0₂)

SECTIONAL PROFILE



ALL DIMENSIONS IN MILLIMETRES

DIMENSION	IEC		EVER BEADY	
	MAX	MIN	MAX	MIN
A/B	30.2	28.0	29.35	28.65
F	4.5		4.5	
G		0.3		0.53
Ø	12.0	10.7	11.95	11.43

nominal open circuit voltage 1.5V maximum open circuit voltage 1.6V average weight 9g immersed volume 33ml nominal capacity 0.9Ah (300 Ω for 12 h/d to 0.9V) recommended storage conditions 20° \pm 5°C, not more than 65% R.H.

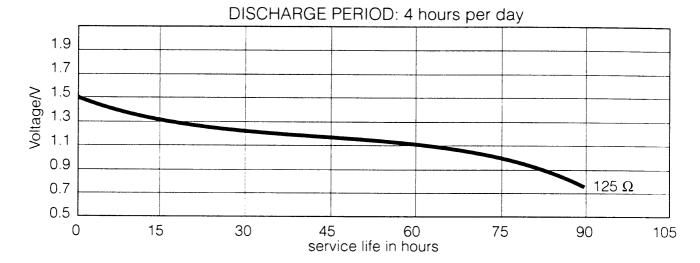
TYPICAL SERVICE LIFE AT 20° C

Applications	Test Conditions	Standard	Typical service life	
hearing aid	300 Ω 12 h/d to 0.9V	_	220h 82h	
radio	125 Ω 4 h/d to to 0.9V			
calculator	30 Ω 1 h/d to 0.9V	_	18h	
flashlight	5.1 Ω 5 m/d to 0.9V	_	132mins	

Recommended shelf life up to 5 years.

TYPICAL PERFORMANCE DATA AT 20° C INITIAL DISCHARGE CURVES

The voltage shown is plotted from the on-load voltage obtained at the end of each day's discharge



Farnell Order Code 279249, 279250, 279262, 279274, 279286, 279298 Data Sheet No CSH00162

COSHH REGULATIONS 1988

Ever Ready Batteries are designed and manufactured, so far as is reasonably practicable, so as to be safe and without risk to health when properly used.

Supplied as sealed units they represent no chemical hazard in the sense of the Control of Substances Hazardous to Health (COSHH) Regulations.

Chemical hazard can however arise if batteries are misused or abused when leakage or, in extreme cases, fire or explosion may occur.

In order to avoid potential problems the Battery Safety Guidelines (copy attached) should be observed on storage, use and disposal.

Detailed Chemical Hazard information for each battery type is attached (Tables 1-7). The following is a summary of the more likely hazards in practice.

1. ZINC CARBON BATTERIES (Ever Ready Blue Range)

The main chemical hazard arises if the battery leaks or vents. The electrolyte is a concentrated solution of zinc chloride and ammonium chloride in water. The material is acidic, corrosive and will cause burns to skin. The electrolyte is also harmful if it enters the eyes. If the user comes into contact with the electrolyte then the part affected should be washed immediately with water. If the material enters the eye medical attention should be sought without delay.

The Cathode mix is corrosive and contains manganese dioxide which is toxic if ingested. Medical attention should be sought if ingestion is thought to have arisen.

2. ZINC CHLORIDE BATTERIES (Ever Ready Silver Seal Range)

The main chemical hazard arises if the battery leaks or vents. The electrolyte is a concentrated solution of zinc chloride in water. This material is acidic, corrosive and will cause burns to skin. The electrolyte is also harmful if it enters the eyes. If the user comes into contact with zinc chloride then the part affected should be washed immediately with water. If the material enters the eye medical attention should be sought without delay.

The cathode mix is corrosive and contains manganese dioxide which is toxic if ingested. Medical attention should be sought if ingestion is thought to have arisen.

3. ALKALINE MANGANESE BATTERIES (Ever Ready Energizer Range)

The main hazard arises if the battery leaks or vents. The electrolyte is strongly alkaline 34-38% w/w potassium hydroxide which is highly corrosive. It will cause burns to skin externally (or internally). Potassium hydroxide is exceedingly harmful if allowed to enter they eyes. Anyone coming into contact with potassium hydroxide should wash with copious amounts of water. Tissue damage is not usually apparent until several hours after exposure. If the material enters the eyes emergency hospital treatment should be sought without delay.

Alkaline manganese cells contain amalgamated zinc powder (1.5%Hg) and manganese dioxide. Both these substances are toxic by ingestion.

4. BUTTON CELLS

Any type of button cell is hazardous if swallowed. If this arises immediate medical attention should be sought. Surgical removal of the battery may be necessary.

The chemical hazard depends on the system type. If button cells are ingested even the nickel plated case material will dissolve in the stomach acid giving rise to toxic nickel salts. Most button cells contain 34-40% potassium hydroxide solution which is highly corrosive but present in small volume.

Mercuric oxide-zinc button cells are the most hazardous if ingested because they contain approximately 30% by cell weight of highly toxic mercuric oxide powder in the cathode. Other button cells also contain amalgamated zinc powder which may be harmful if ingested.

5. <u>NICKEL-CADMIUM BATTERIES</u>

These batteries contain 30% potassium hydroxide solution which is highly corrosive. Normally this material would only be expelled under overcharge conditions. These batteries also contain cadmium, cadmium hydroxide and nickel hydroxide all of which are toxic. If the user comes into contact with potassium hydroxide then the effected area should be washed with a copious supply of water. Potassium hydroxide is harmful if it enters the eyes.

BATTERY SAFETY CODE

Used correctly, domestic batteries are a safe and dependable source of portable power. Problems can occur if they are misused or abused-resulting in leakage or, in extreme cases, fire or explosion.

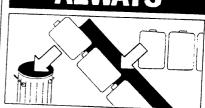
Here are some simple guidelines to safe battery use designed to eliminate any such problems.

ALWAYS



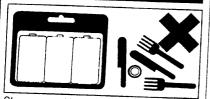
Take care to fit your batteries correctly, observing the *plus* and *minus* marks on the battery and appliance. Incorrect fitting can cause leakage or, in extreme cases, fire or even an explosion.

ALWAYS



Replace the whole set of batteries at one time, taking care not to mix old and new batteries or batteries of different types, since this can result in leakage or, in extreme cases, fire or even an explosion.

ALWAYS



Store unused batteries in their packaging and away from metal objects which may cause a short-circuit resulting in leakage or, in extreme cases, fire or even an explosion.

ALWAYS



Remove dead batteries from equipment and all batteries from equipment you know you are not going to use for a long time. Otherwise the batteries may leak and cause damage.

NEVER



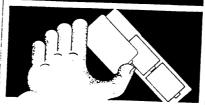
Never dispose of batteries in fire as this can cause them to explode. Please put dead batteries in with the normal household waste.

NEVER



Never attempt to recharge ordinary batteries, either in a charger or by applying heat to them. They may leak, cause fire or even explode. There are special rechargeable batteries which are clearly marked as such.

ALWAYS



Supervise children if they are replacing batteries themselves in order to ensure these guidelines are followed.

ALWAYS



Remember that small button cell batteries such as used in some hearing aids, toys, games and other appliances, are easily swallowed by young children and this can be dangerous.

ALWAYS



Keep loose button cells away from young children.

ALWAYS



Make sure battery compartments are secure

ALWAYS



Seek medical advice if you believe a cell has been swallowed.



Table 3. <u>Components of Energizer Alkaline Manganese range</u> which could be exposed if cells leaked or disrupted on abuse.

SUBSTANCE	HAZARD LEVEL	TYPE OF HAZARO	
Zinc amalgam powder (Anode) High		Harmful mercury vapour released. Harmful by ingestion. Fire	
Carbonal Call		hazard when wet with electrolyte and exposed to air	
Carbopol Gellant	Low	Low oral toxicity, but is in contact with anode and corrosive electrolyte	
Potassium Hydroxide Solution (electrolyte)	High	Highly corrosive, avoid skin contact. Dangerous if enters eye	
Manganese dioxide	High	Toxic material present in largest quantity. Irritant, oxiding agent (assists fire). Avoid ingestion	
raphite	Low	Low toxicity but is mixed with harmful manganese dioxide	
eparator nylon/cellophane laminate		- I STATE OF THE S	
user ny fony ce riophane laminate	LOW	Low toxicity but wet with corrosive potassium hydroxide	
ickel plated steel container/ op cap	LOW	Corroded surfaces would lead to toxic nickel salts avoid	
ass current collector nail	Tok	Low inherent hazard, but nail after contact with amalgam would	